

**Title** A cryoprotective and cold-adapted 1,3- $\beta$ -endoglucanase from cherimoya (*Annona cherimola*) fruit

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### Abstract

A 1,3- $\beta$ -glucanase with potent cryoprotective activity was purified to homogeneity from the mesocarp of CO<sub>2</sub>-treated cherimoyafruit (*Annona cherimola* Mill.) stored at low temperature using anion exchange and chromatofocusing chromatography. This protein was characterized as a glycosylated endo-1,3- $\beta$ -glucanase with a  $M_r$  of 22.07 kDa and a  $pI$  of 5.25. The hydrolase was active and stable in a broad acidic pH range and it exhibited maximum activity at pH 5.0. It had a low optimum temperature of 35 °C and it retained 40% maximum activity at 5 °C. The purified 1,3- $\beta$ -glucanase was relatively heat unstable and its activity declined progressively at temperatures above 50 °C. Kinetic studies revealed low  $k_{cat}$  ( $3.10 \pm 0.04 \text{ s}^{-1}$ ) and  $K_m$  ( $0.32 \pm 0.03 \text{ mg ml}^{-1}$ ) values, reflecting the intermediate efficiency of the protein in hydrolyzing laminarin. Moreover, a thermodynamic characterization revealed that the purified enzyme displayed a high  $k_{cat}$  at both 37 and 5 °C, and a low  $E_a$  ( $6.99 \text{ kJ mol}^{-1}$ ) within this range of temperatures. *In vitro* functional studies indicated that the purified 1,3- $\beta$ -glucanase had no inhibitory effects on *Botrytis cinerea* hyphal growth and no antifreeze activity, as determined by thermal hysteresis analysis using differential scanning calorimetry. However, a strong cryoprotective activity was observed against freeze-thaw inactivation of lactate dehydrogenase. Indeed, the PD<sub>50</sub> was  $8.7 \text{ } \mu\text{g ml}^{-1}$  (394 nM), 9.2-fold higher (3.1 on a molar basis) than that of the cryoprotective protein BSA. Together with the observed accumulation of glycine-betaine in CO<sub>2</sub>-treated cherimoya tissues, these results suggest that 1,3- $\beta$ -glucanase could be functionally implicated in low temperature-defense mechanism activated by CO<sub>2</sub>.